## SUBCHAPTER D—WATER PROGRAMS (CONTINUED)

## PART 136—GUIDELINES ESTAB-LISHING TEST PROCEDURES FOR THE ANALYSIS OF POLLUTANTS

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AUTHORITY: Secs. 301, 304(h), 307 and 501(a), Pub. L. 95–217, 91 Stat. 1566, et seq. (33 U.S.C. 1251, et seq.) (the Federal Water Pollution Control Act Amendments of 1972 as amended by the Clean Water Act of 1977).

## § 136.1 Applicability.

The procedures prescribed herein shall, except as noted in §136.5, be used to perform the measurements indicated whenever the waste constituent specified is required to be measured for:

- (a) An application submitted to the Administrator, or to a State having an approved NPDES program for a permit under section 402 of the Clean Water Act of 1977, as amended (CWA), and/or to reports required to be submitted under NPDES permits or other requests for quantitative or qualitative effluent data under parts 122 to 125 of title 40, and,
- (b) Reports required to be submitted by discharges under the NPDES established by parts 124 and 125 of this chapter, and,
- (c) Certifications issued by States pursuant to section 401 of the CWA, as amended.

[38 FR 28758, Oct. 16, 1973, as amended at 49 FR 43250, Oct. 26, 1984]

#### § 136.2 Definitions.

As used in this part, the term:

- (a) Act means the Clean Water Act of 1977, Pub. L. 95–217, 91 Stat. 1566, et seq. (33 U.S.C. 1251 et seq.) (The Federal Water Pollution Control Act Amendments of 1972 as amended by the Clean Water Act of 1977).
- (b) Administrator means the Administrator of the U.S. Environmental Protection Agency.
- (c) Regional Administrator means one of the EPA Regional Administrators.
- (d) *Director* means the Director of the State Agency authorized to carry out an approved National Pollutant Discharge Elimination System Program under section 402 of the Act.
- (e) National Pollutant Discharge Elimination System (NPDES) means the national system for the issuance of permits under section 402 of the Act and includes any State or interstate program which has been approved by the Administrator, in whole or in part, pursuant to section 402 of the Act.
- (f) Detection limit means the minimum concentration of an analyte (substance) that can be measured and reported with a 99% confidence that the analyte concentration is greater than zero as determined by the procedure set forth at appendix B of this part.

[38 FR 28758, Oct. 16, 1973, as amended at 49 FR 43250, Oct. 26, 1984]

# §136.3 Identification of test procedures.

(a) Parameters or pollutants, for which methods are approved, are listed together with test procedure descriptions and references in Tables IA, IB, IC, ID, IE, and IF. The full text of the referenced test procedures are incorporated by reference into Tables IA, IB, IC, ID, IE, and IF. The references and the sources which are available are given in paragraph (b) of this section. These test procedures are incorporated as they exist on the day of approval and a notice of any change in these test procedures will be published in the FEDERAL REGISTER. The discharge parameter values for which reports are required must be determined by one of

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the standard analytical test procedures incorporated by reference and described in Tables IA, IB, IC, ID, IE, and IF, or by any alternate test procedure which has been approved by the Administrator under the provisions of paragraph (d) of this section and §§136.4 and 136.5. Under certain circumstances (paragraph (b) or (c) of this section or 40 CFR 401.13) other test procedures

may be used that may be more advantageous when such other test procedures have been previously approved by the Regional Administrator of the Region in which the discharge will occur, and providing the Director of the State in which such discharge will occur does not object to the use of such alternate test procedure.

## TABLE IA.—LIST OF APPROVED BIOLOGICAL METHODS

Parameter and units	Method <sup>1</sup>	EPA	Standard methods, 18th Ed.	ASTM	USGS
Bacteria:					
1. Coliform (fecal), number	Most Probable Number (MPN), 5 tube	p. 132 <sup>3</sup>	9221C E4		
per 100 mL.	3 dilution, or Membrane filter (MF) <sup>2</sup> , single step	p. 124 <sup>3</sup>	9222D4		B-0050-855
2. Coliform (fecal) in presence	MPN, 5 tube, 3 dilution, or	p. 132 <sup>3</sup>	9221C E <sup>4</sup>		
of chlorine, number per 100	MF, single step 6	p. 124 <sup>3</sup>	9222D4		
mL.					
<ol><li>Coliform (total), number per</li></ol>	MPN, 5 tube, 3 dilution, or	p. 114 <sup>3</sup>	9221B4		
100 mL.	MF <sup>2</sup> single step or two step	p. 108 <sup>3</sup>	9222B4		B-0025-855
<ol><li>Coliform (total), in presence</li></ol>	MPN, 5 tube, 3 dilution, or	p. 114 <sup>3</sup>	9221B4		
of chlorine, number per 100	MF 2 with enrichment	p. 111 <sup>3</sup>	9222(B+B.5c)4		
mL.					
<ol><li>Fecal streptococci, number</li></ol>		p. 139 <sup>3</sup>	9230B4		
per 100 mL.	MF <sup>2</sup> , or	p. 136 <sup>3</sup>	9230C4		B-0055-85 5
	Plate count	p. 143 <sup>3</sup>			
Aquatic Toxicity:					
6. Toxicity, acute, fresh water	Daphnia, Ceriodaphnia, Fathead Minnow, Rainbow Trout, Brook	Sec. 9 <sup>7</sup>			
organisms, LC50, percent	Trout, or Bannerfish Shiner mortality.				
effluent.					
7. Toxicity, acute, estuarine	Mysid, Sheepshead Minnow, or Menidia spp. mortality	Sec. 97			
and marine organisms,					
LC50, percent effluent.		4000 00			
8. Toxicity, chronic, fresh		1000.08			
water organisms, NOEC or	Fathead minnow embryo-larval survival and teratogenicity	1001.08			
IC25, percent effluent.	Ceriodaphnia survival and reproduction	1002.08			
	Selenastrum growth	1002.08			
9. Toxicity, chronic, estuarine	Sheepshead minnow larval survival and growth	1003.09			
	Sheepshead minnow embryo-larval survival and teratogenicity	1004.09			
and marine organisms, NOEC or IC25, percent ef-	Menidia beryllina larval and growth	1005.0			
fluent.	Mysidopsis bahia survival, growth, and fecundity	1006.09			
nuont.	Arbacia punctulata fertilization	1007.09			
	Champia parvula reproduction	1007.09			
	Onampia parvaia reproduction	1009.09			

<sup>&</sup>lt;sup>1</sup> The method must be specified when results are reported.

<sup>&</sup>lt;sup>2</sup>A 0.45 um membrane filter (MF) or other pore size certified by the manufacturer to fully retain organisms to be cultivated and to be free of extractables which could interfere with their

<sup>4 0.45</sup> um membrane filter (MF) or other pore size certified by the manufacturer to fully retain organisms to be cultivated and to be free of extractables which could interfere with their growth.

3 USEPA. 1978. Microbiological Methods for Monitoring the Environment, Water, and Wastes. Environmental Monitoring and Support Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. EPA/600/8–78/017.

4 APHA. 1992. Standard Methods for the Examination of Water and Wastewater. American Public Health Association. 18th Edition. Amer. Publ. HIth. Assoc., Washington, DC.

5 USGS. 1989. U.S. Geological Survey Techniques of Water-Resources Investigations, Book 5, Laboratory Analysis, Chapter A4, Methods for Collection and Analysis of Aquatic Biological and Microbiological Samples, U.S. Geological Survey, U.S. Department of Interior, Reston, Virginia.

<sup>&</sup>lt;sup>6</sup>Because the MF technique usually yields low and variable recovery from chlorinated wastewaters, the Most Probable Number method will be required to resolve any controversies.

<sup>7</sup>USEPA. 1993. Methods for Measuring the Acute Toxicity of Effluents to Freshwater and Marine Organisms. Fourth Edition. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. August 1993, EPA/600/4–90/027F.

<sup>®</sup>USEPA. 1994. Short-term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Freshwater Organisms. Third Edition. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency USEPA. 1994, Cincinnati, Ohio (July 1994, EPA/600/4–91/002).

<sup>®</sup>Short-term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Marine and Estuarine Organisms. Second Edition. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio (July 1994, EPA/600/4–91/003). These methods do not apply to marine waters of the Pacific Ocean.

#### TABLE IB.—LIST OF APPROVED INORGANIC TEST PROCEDURES

	Reference (method number or page)						
Parameter, units and method	EPA 1,35	STD methods 18th ed.	ASTM	USGS <sup>2</sup>	Othe		
. Acidity, as CaCO <sub>3</sub> , mg/L:							
Electrometric endpoint or phenolphthalein endpoint 2. Alkalinity, as CaCO <sub>3</sub> , mg/L:	305.1	2310 B(4a)	D1067–92				
Electrometric or Colorimetric titration to pH 4.5,	310.1	2320 B	D1067–92	I–1030–85	973.43.3		
manual or automated.	310.2			I-2030-85			
. Aluminum—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:							
AA direct aspiration 36	202.1	3111 D		I-3051-85			
AA furnace	202.2			. 555. 55			
Inductively Coupled Plasma/Atomic Emission Spec-	5 200.7						
trometry (ICP/AES) 36.	200.7				1		
Direct Current Plasma (DCP) <sup>36</sup>			D4190-82(88)		Note 34.		
Colorimetric (Eriochrome cyanine R)		3500-ALD	D4190-02(00)		11016 34.		
. Ammonia (as N), mg/L:		3500-ALD					
Manual, distillation (at pH 9.5),6 followed by	350.2	4500-NH <sub>3</sub> B			973.49.3		
				I–3520–85			
Nesslerization	350.2		D1426-93(A)	I-3520-85	973.49. <sup>3</sup>		
Titration	350.2		D. 100 00/D)				
Electrode	350.3						
Automated phenate, or	350.1	4500–NH <sub>3</sub> H		I-4523-85			
Automated electrode					Note 7.		
. Antimony-Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:		=					
AA direct aspiration 36		3111 B					
AA furnace		3113 B					
ICP/AES 36	5 200.7	3120 B					
i. Arsenic-Total,4 mg/L:							
Digestion 4 followed by	206.5						
AA gaseous hydride	206.3	3114 B 4.d	D2972-93(B)	I-3062-85			
AA furnace	206.2	3113 B	D2972-93(C)				
ICP/AES, <sup>36</sup> or	5 200.7	3120 B					
Colorimetric (SDDC)	206.4	3500-As C	D2972-93(A)	I-3060-85			
'. Barium—Total,4 mg/L; Digestion 4 followed by:							
AA direct aspiration 36	208.1	3111 D		I-3084-85	1		
AA furnace	208.2		D4382-91		1		
ICP/AES 36	5 200.7						
DCP 36	200.7	0120 B			Note 34.		
B. Beryllium—Total, 4 mg/L; Digestion 4 followed by:					1.3.00		
AA direct aspiration	210 1	3111 D	D3645-93(88)(A)	I-3095-85	1		
AA furnace		3113 B		. 5555 55	1		
ICP/AES		3120 B	D3043-33(00)(B)				
DCP, or			D4100 92(99)		Note 24		

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Colorimetric (aluminon)		3500-Be D			
9. Biochemical oxygen demand (BOD <sub>5</sub> ), mg/L:					
Dissolved Oxygen Depletion	405.1	5210 B		I–1578–78 <sup>8</sup>	973.44, <sup>3</sup> p. 17. <sup>9</sup>
Colorimetric (curcumin)	212.3	4500-B B		1 2442 05	
	5 200.7			1-3112-65	
ICP/AES, or			D.1100 00/00)		
DCP			D4190-82(88)		Note 34
11. Bromide, mg/L:					
Titrimetric	320.1		D1246-82(88)(C)	I–1125–85	p. S44. <sup>10</sup>
<ol> <li>Cadmium—Total,<sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:</li> </ol>					
AA direct aspiration 36	213.1	3111 B or C	D3557-90(A or B)	I-3135-85 or I-3136-85	974.27,3 p. 37.9
AA furnace	213.2	3113 B	D3557-90(D)		
ICP/AES 36	5 200.7	3120 B		I-1472-85	
DCP 36			D4190-82(88)		Note 34.
Voltametry, <sup>11</sup> or			D3557-90(C)		
Colorimetric (Dithizone)			20007 00(0)		
13. Calcium—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:		3300-Cd B			
	045.4	0444 B	D544 00/D)	1 0450 05	
AA direct aspiration	215.1		D511–93(B)	I-3152-85	
ICP/AES	5 200.7				
DCP, or					Note 34.
Titrimetric (EDTA)	215.2	3500-Ca D	D511-93(A)		
4. Carbonaceous biochemical oxygen demand					
(CBOD <sub>5</sub> ), mg/L <sup>12</sup> :					
Dissolved Oxygen Depletion with nitrification inhibitor.		5210 B			
15. Chemical oxygen demand (COD), mg/L; Titrimetric,	410.1	5220 C	D1252-88(A)	I-3560-85	973.46,3 p. 17.9
or.	410.2	0220 0	21202 00(1)	I–3562–85	0.0o, p
01.	410.3			1 0002 00	
Spectrophotometric, manual or automated	410.4	5220 D	D1252-88(B)	I-3561-85	Notes 13 or 14.
16. Chloride, mg/L:	410.4	3220 D	D1232-60(B)	1-3301-83	110165 13 01 14.
Titrimetric (silver nitrate) or		4500-CI- B	D512–89(B)	1 4402 05	
					070 54 3
(Mercuric nitrate)	325.3	4500-CI- C			973.51. <sup>3</sup>
Colorimetric, manual or					
Automated (Ferricyanide)	325.1 or	4500-CI-E		I-2187-85	
	325.2				
<ol> <li>Chlorine—Total residual, mg/L; Titrimetric:</li> </ol>					
Amperometric direct	330.1	4500-CI D	D1253-86(92)		
lodometric direct	330.3	4500-CI B	` ′		
Back titration ether end- point 15 or	330.2				
DPD-FAS	330.4				
Spectrophotometric, DPD	330.5				
					Note 40
Or Electrode					Note 16.
18. Chromium VI dissolved, mg/L; 0.45 micron filtration					
followed by:					
AA chelation-extraction or	218.4			I-1232-85	
Colorimetric (Diphenylcarbazide)		3500-Cr D	D1687–92(A)	I-1230-85	
<ol> <li>Chromium—Total,<sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:</li> </ol>					
AA direct aspiration 36	218.1	3111 B	D1687-92(B)	I-3236-85	974.27. <sup>3</sup>
					1 -
AA chelation-extraction	218.3	3111 C			

	Reference (method number or page)						
Parameter, units and method	EPA 1,35	STD methods 18th ed.	ASTM	USGS <sup>2</sup>	Other		
ICP/AES 36	5 200.7	3120 B					
DCP, 36 or			D4190-82(88)		Note 34.		
Colorimetric (Diphenylcarbazide)  0. Cobalt—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:		3500-Cr D					
AA direct aspiration	219.1	3111 B or C	D3558-90(A or B)	I-3239-85	p. 37.9		
AA furnace	219.2	3113 B	D3558-90(C)				
ICP/AES	5 200.7	3120 B	1				
DCP			D4190-82(88)		Note 34.		
Color platinum cobalt units or dominant wavelength, hue, luminance purity:							
Colorimetric (ADMI), or	110.1	2120 E			Note 18.		
(Platinum cobalt), or	110.2	2120 B		I-1250-85			
Spectrophotometric	110.3	2120 C					
2. Copper—Total,4 mg/L; Digestion 4 followed by:							
AA direct aspiration 36	220.1	3111 B or C	D1688–90(A or B)	I-3270-85 or I3271-85	974.27 <sup>3</sup> p. 37.		
AA furnace	220.2	3113 B	D1688-90(C)				
ICP/AES 36	5 200.7	3120 B					
DCP 36 or			D4190-82(88)		Note 34.		
Colorimetric (Neocuproine) or		3500-Cu D	2 1100 02(00)		11010 0 11		
(Bicinchoninate)		Or E			Note 19.		
3. Cyanide—Total, mg/L:		0.2					
Manual distillation with MgCl <sub>2</sub> followed by		4500-CN C	D2036-91(A)				
Titrimetric, or		4500-CN D			p. 22.9		
Spectrophotometric, manual or	31 335.2	4500-CN E		I-3300-85	P. 22.		
Automated 20	31 335.3	1000 014 2	D2000 01(71)	1 0000 00			
4. Available Cyanide, mg/L	000.0						
Cyanide amenable to chlorination (CATC), Manual	335.1	4500-CN G	D2036-91(B)				
distillation with MgCl <sub>2</sub> followed by titrimetry or spectrophotometry.	000.1	4000 014 0	52000 51(B):				
Flow injection and ligand exchange, followed by					44 OIA-1677		
amperometry.							
5. Fluoride—Total, mg/L:							
Manual distillation 6 followed by		4500-F B					
Electrode, manual or	340.2	4500-F C	D1179-93(B)				
Automated				I-4327-85			
Colorimetric (SPADNS)	340.1	4500-F D	D1179-93(A)				
Or Automated complexone	340.3						
6. Gold—Total,4 mg/L; Digestion 4 followed by:							
AA direct aspiration	231.1	3111 B					
AA furnace, or	231.2						
DCP					Note 34.		
7. Hardness—Total, as CaCO <sub>3</sub> , mg/L							
Automated colorimetric,	130.1						

TABLE IB.—LIST OF APPROVED INORGANIC TEST PROCEDURES—Continued

Titrimetric (EDTA), or Ca plus Mg as their carbon- ates, by inductively coupled plasma or AA direct aspiration. (See Parameters 13 and 33).		2340 B or C	D1126-86(92)	I–1338–85	973.52B. <sup>3</sup>
28. Hydrogen ion (pH), pH units  Electrometric measurement, or	150.1	4500–H= B	D1293-84(90)(A or B)	I–1586–85	973.41.3
Automated electrode		4300-н- В	D1293-84(90)(A 01 B)		Note 21.
29. Iridium—Total, 4 mg/L; Digestion 4 followed by:					
AA direct aspiration or	235.1	3111 B			
AA furnace	235.2				
30. Iron—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:		<u>_</u> _			
AA direct aspiration 36		3111 B or C	D1068–90(A or B)	I-3381-85	974.27.3
AA furnace		3113 B 3120 B	D1068-90(C)		
ICP/AES 36			D4400 92(99)		Note 34.
DCP <sup>36</sup> or  Colorimetric (Phenanthroline)		3500–Fe D	D4190-82(88) D1068-90(D)		Note 34.
31. Kjeldahl Nitrogen—Total, (as N), mg/L:		3500-Fe D	D1068-90(D)		Note 22.
Digestion and distillation followed by:	351.3	4500-NH <sub>3</sub> B or C	D3590-89(A)		
Titration		4500-NH <sub>3</sub> E	D3590-89(A)		973.48 <sub>3</sub> .
Nesslerization		4500-NH <sub>3</sub> C	D3590-89(A)		313.403.
Electrode		4500–NH <sub>3</sub> F or G	B3330-03(A)		
Automated phenate colorimetric		4500-141131 01 0		I–4551–78 <sub>8</sub>	
Semi-automated block digester colorimetric			D3590-89(B)	1-4331-708	
Manual or block digester potentiometric			D3590-89(A)		
Block Digester, followed by:.			20000 00(7.1)		
Auto distillation and Titration, or					Note 39.
Nesslerization					Note 40.
Flow injection gas diffusion					Note 41.
32. Lead—Total,4 mg/L; Digestion4 followed by:					
AA direct aspiration 36	239 1	3111 B or C	D3559-90(A or B)	I-3399-85	974.27.3
AA furnace			D3559-90(D)		0
ICP/AES 36		3120 B			
DCP 36			D4190-82(88)		Note 34.
Voltametry 11 or			D3559-90(C)		
Colorimetric (Dithizone)		3500-Pb D	. ,		
33. Magnesium—Total,4 mg/L; Digestion4 followed by:					
AA direct aspiration	242.1	3111 B	D511–93(B)	I-3447-85	974.27.3
ICP/AES	5 200.7	3120 B			
DCP, or					Note 34.
Gravimetric		3500–Mg D			
34. Manganese—Total,4 mg/L; Digestion4 followed by:					
AA direct aspiration 36		3111 B	D858–90(A or B)	I-3454-85	974.27. <sup>3</sup>
AA furnace			D858-90(C)		
ICP/AES 36		3120 B			
DCP 36 or			D4190-82(88)		Note 34.
Colorimetric (Persulfate), or		3500–Mn D			920.203.3
(Periodate)					Note 23.
35. Mercury—Total, <sup>4</sup> mg/L:	045 4	0440 B	D0000 04	1 0400 05	2077.00
Cold vapor, manual, or		3112 B	D3223–91	I-3462-85	3977.22
Automated	245.2		l	I	I

# TABLE IB.—LIST OF APPROVED INORGANIC TEST PROCEDURES—Continued

	Reference (method number or page)						
Parameter, units and method	EPA 1,35	STD methods 18th ed.	ASTM	USGS <sup>2</sup>	Other		
Oxidation, purge and trap, and cold vapor atomic	<sup>43</sup> 1631						
fluorescence spectrometry (ng/L).							
36. Molybdenum—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:							
AA direct aspiration	246.1	3111 D		I-3490-85			
AA furnace		3113 B					
ICP/AES	5 200.7	3120 B					
DCP					Note 34.		
37. Nickel—Total,4 mg/L; Digestion 4 followed by:							
AA direct aspiration 36	249 1	3111 B or C	D1886–90(A or B)	I-3499-85			
AA furnace		3113 B	D1886–90(C)	. 5.55 55			
ICP/AES 36	5 200.7	3120 B	D1000-30(0)				
DCP 36, or	200.7		D4190-82(88)		Note 34.		
Colorimetric (heptoxime)		3500-Ni D	D4130-02(00)		Note 54.		
38. Nitrate (as N), mg/L:		3300-141 B					
Colorimetric (Brucine sulfate), or Nitrate-nitrite N	352.1				973.50, <sup>3</sup> 419 D, <sup>17</sup> p.		
minus Nitrite N (See parameters 39 and 40).	332.1				28.9		
39. Nitrate-nitrite (as N), mg/L:					26.0		
Cadmium reduction. Manual or	252.2	4500-NO <sub>3</sub> - E	D3867-90(B)				
Automated, or	353.2	4500-NO <sub>3</sub> - E	D3867-90(B)	I-4545-85			
	353.2 353.1		D3867-90(A)	1-4545-65			
Automated hydrazine	333.1	4500–NO <sub>3</sub> – H					
40. Nitrite (as N), mg/L; Spectrophotometric:	0544	4500–NO <sub>2</sub> – B			Nata OF		
Manual or	354.1	<del>-</del>		L 4540 05	Note 25.		
Automated (Diazotization)	442.4	FEO. D38		I-4540-85			
41. Oil and grease—Total recoverable, mg/L:	413.1	5520 B <sup>38</sup> .					
Gravimetric (extraction)	4004 B						
Oil and grease and non-polar material, mg/L:	1664, Rev.						
Hexane extractable material (HEM): n-Hexane	Α						
extraction and gravimetry <sup>42</sup> .							
Silica gel treated HEM (SGT-HEM): Silica gel treat-	1664, Rev.						
ment and gravimetry <sup>42</sup> .	Α						
42. Organic carbon—Total (TOC), mg/L:							
Combustion or oxidation	415.1	5310 B, C, or D	D2579–93 (A or B)		973.47, <sup>3</sup> p. 14. <sup>24</sup>		
43. Organic nitrogen (as N), mg/L:							
Total Kjeldahl N (Parameter 31) minus ammonia N							
(Parameter 4)							
44. Orthophosphate (as P), mg/L; Ascorbic acid method:							
Automated, or	365.1	4500–P F		I–4601–85	973.56. <sup>3</sup>		
Manual single reagent	365.2	4500-P E	D515-88(A)		973.55 <sup>3</sup> .		
Manual two reagent	365.3						
45. Osmium—Total 4, mg/L; Digestion 4 followed by:							
AA direct aspiration, or	252.1	3111 D					
AA furnace	252.2						
46. Oxygen, dissolved, mg/L:							
Winkler (Azide modification), or	360.2	4500-O C	D888-92(A)	I-1575-78 8	973.45B.3		

Electrode	360.1	4500–O G	D888–92(B)	I-1576-78 8	
47. Palladium—Total,4 mg/L; Digestion4 followed by:		=			
AA direct aspiration, or	253.1	3111 B			p. S27.10
AA furnace	253.2				p. S28.10
DCP					Note 34.
48. Phenols, mg/L:					
Manual distillation 26	420.1				Note 27.
Followed by:					
Colorimetric (4AAP) manual, or	420.1				Note 27.
Automated 19	420.2				
49. Phosphorus (elemental), mg/L:					
, ,					Note 28.
50. Phosphorus—Total, mg/L:					11010 20.
Persulfate digestion followed by	365.2	4500-P B,5			973.55.3
Manual or	365.2 or	4500-P E			913.33.
Ivialiual 01		4500-P E	D313-66(A)		
	365.3	4500 B 5		1 4000 05	070 500
Automated ascorbic acid reduction	365.1	4500-P F		I–4600–85	973.56. <sup>3</sup>
Semi-automated block digestor	365.4		D515-88(B)		
51. Platinum—Total,4 mg/L; Digestion 4 followed by:					
AA direct aspiration	255.1	3111 B			
AA furnace	255.2				
DCP					Note 34.
52. Potassium—Total,4 mg/L; Digestion 4 followed by:					
AA direct aspiration	258.1	3111 B		I-3630-85	973.53. <sup>3</sup>
ICP/AES		3120 B		. 5555 55	0.000
Flame photometric, or		3500-K D			
Colorimetric		3300-10 5			317 B. <sup>17</sup>
53. Residue—Total, mg/L:					317 D.
Gravimetric, 103–105°	100.0	2540 B		1 2750 05	
	160.3	2540 B		I-3750-85	
54. Residue—filterable, mg/L:	400.4	0540.0		1 1750 05	
Gravimetric, 180°	160.1	2540 C		I–1750–85	
55. Residue—nonfilterable (TSS), mg/L:					
Gravimetric, 103-105° post washing of residue	160.2	2540 D		I-3765-85	
56. Residue—settleable, mg/L:					
Volumetric, (Imhoff cone), or gravimetric	160.5	2540 F			
57. Residue—Volatile, mg/L:					
Gravimetric, 550°	160.4			I-3753-85	
58. Rhodium—Total,4 mg/L; Digestion 4 followed by:					
AA direct aspiration, or	265.1	3111 B			
AA furnace	265.2	0111   12			
59. Ruthenium—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:	200.2				
	207.4	3111 B			
AA direct aspiration, or	267.1 267.2	3111 B			
AA furnace	201.2				
60. Selenium—Total,4 mg/L; Digestion4 followed by:	070 -	0440.5	Doors co(D)		
AA furnace		3113 B	D3859-93(B)		
ICP/AES, <sup>36</sup> or	5 200.7				
AA gaseous hydride		3114 B	D3859-93(A)	I-3667-85	
61. Silica 37—Dissolved, mg/L; 0.45 micron filtration fol-					
lowed by:					
Colorimetric, Manual or	370.1	4500-Si D	D859-88	I-1700-85	

	Reference (method number or page)					
Parameter, units and method	EPA 1,35	STD methods 18th ed.	ASTM	USGS <sup>2</sup>	Other	
Automated (Molybdosilicate), or				1–2700–85		
ICP	5 200.7	3120 B				
2. Silver—Total, 4 mg/L; Digestion 4, 29 followed by:						
AA direct aspiration	272.1	3111 B or C		I-3720-85	974.27,3 p. 37	
AA furnace	272.2	3113 B				
ICP/AES	5 200.7	3120 B				
DCP					Note 34.	
. Sodium—Total,4 mg/L; Digestion4 followed by:						
AA direct aspiration	273.1	3111 B		I-3735-85	973.54.3	
ICP/AES	5 200.7	3120 B				
DCP, or					Note 34.	
Flame photometric		3500 Na D				
. Specific conductance, micromhos/cm at 25 °C:						
Wheatstone bridge	120.1	2510 B	D1125–91(A)	I-1780-85	973.40. <sup>3</sup>	
. Sulfate (as SO <sub>4</sub> ), mg/L:			1			
Automated colorimetric (barium chloranilate)	375.1					
Gravimetric	375.3	4500-SO <sub>4</sub> -2 C or D			925.54.3	
Turbidimetric, or	375.4		D516–90		426C. <sup>30</sup>	
. Sulfide (as S), mg/L:						
Titrimetric (iodine), or	376.1	4500-S-2E		I-3840-85		
Colorimetric (methylene blue)	376.2	4500-S-2D				
. Sulfite (as SO <sub>3</sub> ), mg/L:						
Titrimetric (iodine-iodate)	377.1	4500-SO <sub>3</sub> -2 B				
Surfactants, mg/L:						
Colorimetric (methylene blue)	425.1	5540 C	D2330–88			
. Temperature, °C:						
Thermometric	170.1	2550 B			Note 32.	
. Thallium—Total,4 mg/L; Digestion 4 followed by:						
AA direct aspiration	279.1	3111 B				
AA furnace	279.2					
ICP/AES, or	5 200.7	3120 B				
. Tin—Total,4 mg/L; Digestion 4 followed by:						
AA direct aspiration	282.1	3111 B		I-3850-78 8		
AA furnace, or	282.2	3113 B				
ICP/AES	5 200.7					
. Titanium—Total,4 mg/L; Digestion 4 followed by:						
AA direct aspiration	283.1	3111 D				
AA furnace	283.2					
DCP	200.2				Note 34.	
. Turbidity, NTU:						
Nephelometric	180.1	2130 B	D1889–88(A)	I–3860–85		
. Vanadium—Total,4 mg/L; Digestion4 followed by:			- 1500 000,			
AA direct aspiration	286 1	3111 D				
AA furnace			D3373_93			

TABLE IB.—LIST OF APPROVED INORGANIC TEST PROCEDURES—Continued

ICP/AES		3120 B	D4190-82(88)		Note 34.
DCP, or  Colorimetric (Gallic acid)			D4190-62(66)		Note 34.
75. Zinc—Total, <sup>4</sup> mg/L; Digestion <sup>4</sup> followed by:		3300-V B			
AA direct aspiration 36	289.1	3111 B or C	D1691–90 (A or B)	I-3900-85	974.27, <sup>3</sup> p. 37. <sup>9</sup>
AA furnace	289.2				
ICP/AES 36		3120 B			
DCP, <sup>36</sup> or			D4190-82(88)		Note 34.
Colorimetric (Dithizone) or		3500–Zn E			
(Zincon)		3500–Zn F			Note 33.

Table IR Notes

"Methods for Chemical Analysis of Water and Wastes", Environmental Protection Agency, Environmental Monitoring Systems Laboratory-Cincinnati (EMSL-CI), EPA-600/4-79-020, Revised March 1983 and 1979 where applicable.

<sup>2</sup>Fishman, M.J., et al, "Methods for Analysis of Inorganic Substances in Water and Fluvial Sediments," U.S. Department of the Interior, Techniques of Water—Resource Investigations of the U.S. Geological Survey, Denver, CO, Revised 1989, unless otherwise stated.

<sup>3</sup> "Official Methods of Analysis of the Association of Official Analytical Chemists," methods manual, 15th ed. (1990).

<sup>4</sup> For the determination of total metals the sample is not filtered before processing. A digestion procedure is required to solubilize suspended material and to destroy possible organic-metal complexes. Two digestion procedures are given in "Methods for Chemical Analysis of Water and Wastes, 1979 and 1983". One (section 4.1.3), is a vigorous digestion using nitric acid. A less vigorous digestion using nitric and hydrochloric acids (section 4.1.4) is preferred; however, the analyst should be cautioned that this mild digestion may not suffice for all samples types. Particularly, if a colorimetric procedure is to be employed, it is necessary to ensure that all organo-metallic bonds be broken so that the metal is in a reactive state. In those situations, the vigorous digestion is to be preferred making certain that at no time does the sample go to dryness. Samples containing large amounts of organic materials may also benefit by this vigorous digestion, however, vigorous digestion with concentrated nitric acid will convert antimony and tin to insoluble oxides and render them unavailable for analysis. Use of ICP/AES as well as determinations for certain elements such as antimony, arsenic, the noble metals, mercury, selenium, silver, tin, and titanium require a modified sample digestion procedure and in all cases the method write-up should be consulted for specific instructions and/or cautions.

NOTE TO TABLE IB NOTE 4: If the digestion procedure for direct aspiration AA included in one of the other approved references is different than the above, the EPA procedure must be

Dissolved metals are defined as those constituents which will pass through a 0.45 micron membrane filter. Following filtration of the sample, the referenced procedure for total metals must be followed. Sample digestion of the filtrate for dissolved metals (or digestion of the original sample solution for total metals) may be omitted for AA (direct aspiration or graphite furnace) and ICP analyses, provided the sample solution to be analyzed meets the following criteria:

a. has a low COD (<20)

- b. is visibly transparent with a turbidity measurement of 1 NTU or less
- c. is colorless with no perceptible odor, and

d. is of one liquid phase and free of particulate or suspended matter following acidification.

The full text of Method 200.7, "Inductively Coupled Plasma Atomic Emission Spectrometric Method for Trace Element Analysis of Water and Wastes," is given at Appendix C of this Part

6 Manual distillation is not required if comparability data on representative effluent samples are on company file to show that this preliminary distillation step is not necessary: however, manual distillation will be required to resolve any controversies.

7 Ammonia, Automated Electrode Method, Industrial Method Number 379–75 WE, dated February 19, 1976, (Bran & Luebbe (Technicon) Auto Analyzer II, Bran & Luebbe Analyzing Technologies, Inc., Elmsford, NY 10523.

The approved method is that cited in "Methods for Determination of Inorganic Substances in Water and Fluvial Sediments", USGS TWRI, Book 5, Chapter A1 (1979).

American National Standard on Photographic Processing Effluents, Apr. 2, 1975. Available from ANSI, 1430 Broadway, New York, NY 10018.

Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency", Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981).

11 The use of normal and differential pulse voltage ramps to increase sensitivity and resolution is acceptable.

12 Carbonaceous biochemical oxygen demand (ČBOD<sub>5</sub>) must not be confused with the traditional BOD<sub>5</sub> test which measures "total BOD". The addition of the nitrification inhibitor is not a procedural option, but must be included to report the CBODs parameter. A discharger whose permit requires reporting the traditional BODs may not use a nitrification inhibitor in the procedure for reporting the results. Only when a discharger's permit specifically states CBOD<sub>5</sub> is required can the permittee report data using the nitrification inhibitor.

13 OIC Chemical Oxygen Demand Method, Oceanography International Corporation, 1978, 512 West Loop, P.O. Box 2980, College Station, TX 77840.

14 Chemical Oxygen Demand, Method 8000, Hach Handbook of Water Analysis, 1979, Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.

<sup>15</sup> The back titration method will be used to resolve controversy.

16 Orion Research Instruction Manual, Residual Chlorine Electrode Model 97-70, 1977, Orion Research Incorporated, 840 Memorial Drive, Cambridge, MA 02138. The calibration graph for the Orion residual chlorine method must be derived using a reagent blank and three standard solutions, containing 0.2, 1.0, and 5.0 ml 0.00281 N potassium iodate/100 ml solution, respectively.

17 The approved method is that cited in Standard Methods for the Examination of Water and Wastewater, 14th Edition, 1976.

18 National Council of the Paper Industry for Air and Stream Improvement, (Inc.) Technical Bulletin 253, December 1971.

19 Copper, Biocinchoinate Method, Method 8506, Hach Handbook of Water Analysis, 1979, Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.

- 20 After the manual distillation is completed, the autoanalyzer manifolds in EPA Methods 335.3 (cyanide) or 420.2 (phenols) are simplified by connecting the re-sample line directly to the sampler. When using the mainfold setup shown in Method 335.3, the buffer 6.2 should be replaced with the buffer 7.6 found in Method 335.2.

  21 Hydrogen ion (pH) Automated Electrode Method, Industrial Method Number 378–75WA, October 1976, Bran & Luebbe (Technicon) Autoanalyzer II. Bran & Luebbe Analyzing Technicon
- nologies, Inc., Elmsford, NY 10523.
  - <sup>22</sup> Iron, 1,10-Phenanthroline Method, Method 8008, 1980, Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>23</sup> Manganese, Periodate Oxidation Method, Method 8034, Hach Handbook of Wastewater Analysis, 1979, pages 2–113 and 2–117, Hach Chemical Company, Loveland, CO 80537.
- 24 Wershaw, R.L., et al, "Methods for Analysis of Organic Substances in Water," Techniques of Water-Resources Investigation of the U.S. Geological Survey, Book 5, Chapter A3, (1972
- <sup>25</sup> Nitrogen, Nitrite, Method 8507, Hach Chemical Company, P.O. Box 389, Loveland, CO 80537.
- <sup>26</sup> Just prior to distillation, adjust the sulfuric-acid-preserved sample to pH 4 with 1 + 9 NaOH.
- 27 The approved method is cited in Standard Methods for the Examination of Water and Wastewater, 14th Edition. The colorimetric reaction is conducted at a pH of 10.0±0.2. The approved methods are given on pp 576–81 of the 14th Edition: Method 510A for distillation, Method 510B for the manual colorimetric procedure, or Method 510C for the manual spectophotometric procedure.
- <sup>29</sup> R. F. Addison and R.G. Ackman, "Direct Determination of Elemental Phosphorus by Gas-Liquid Chromatography," Journal of Chromatography, vol. 47, No. 3, pp. 421–426, 1970.
  <sup>29</sup> Approved methods for the analysis of silver in industrial wastewaters at concentrations of 1 mg/L and above are inadequate where silver exists as an inorganic halide. Silver halides
- such as the bromide and chloride are relatively insoluble in reagents such as nitric acid but are readily soluble in an aqueous buffer of sodium thiosulfate and sodium hydroxide to pH of 12. Therefore, for levels of silver above 1 mg/L, 20 mL of sample should be diluted to 100 mL by adding 40 mL each of 2 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and NaOH. Standards should be prepared in the same manner. For levels of silver below 1 mg/L the approved method is satisfactory.
- <sup>30</sup>The approved method is that cited in Standard Methods for the Examination of Water and Wastewater, 15th Edition.
- 31 EPA Methods 335.2 and 335.3 require the NaOH absorber solution final concentration to be adjusted to 0.25 N before colorimetric determination of total cyanide.
- 32 Stevens, H.H., Ficke, J.F., and Smoot, G.F., "Water Temperature—Influential Factors, Field Measurement and Data Presentation", Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 1, Chapter D1, 1975.
- 33 Zinc, Zincon Method, Method 8009, Hach Handbook of Water Analysis, 1979, pages 2–231 and 2–333, Hach Chemical Company, Loveland, CO 80537.
  34 "Direct Current Plasma (DCP) Optical Emission Spectrometric Method for Trace Elemental Analysis of Water and Wastes, Method AES0029," 1986—Revised 1991, Fison Instruments, Inc., 32 Commerce Center, Cherry Hill Drive, Danvers, MA 01923.

  35 Precision and recovery statements for the atomic absorption direct aspiration and graphite furnace methods, and for the spectrophotometric SDDC method for arsenic are provided in
- Appendix D of this part titled, "Precision and Recovery Statements for Methods for Measuring Metals".

  36 "Closed Vessel Microwave Digestion of Wastewater Samples for Determination of Metals", CEM Corporation, P.O. Box 200, Matthews, NC 28106–0200, April 16, 1992. Available from
- the CEM Corporation.
  - <sup>37</sup>When determining boron and silica, only plastic, PTFE, or quartz laboratory ware may be used from start until completion of analysis

  - 37 When determining boron and silica, only plastic, PTFE, or quartz laboratory ware may be used from start unit completion or analysis.

    38 Only the trichlorofluoromethane extraction solvent is approved.

    39 Nitrogen, Total Kjeldahl, Method PAI–DK01 (Block Digestion, Steam Distillation, Titrimetric Detection), revised 12/22/94, Perstop Analytical Corporation.

    40 Nitrogen, Total Kjeldahl, Method PAI–DK02 (Block Digestion, Steam Distillation, Colorimetric Detection), revised 12/22/94, Perstop Analytical Corporation.

    41 Nitrogen, Total Kjeldahl, Method PAI–DK03 (Block Digestion, Automated FIA Gas Diffusion), revised 12/22/94, Perstop Analytical Corporation.

    42 Method 1664, Revision A "n-Hexane Extractable Material (HEM; Oil and Grease) and Silica Gel Treated n-Hexane Extractablke Material (SGT-HEM; Non-polar Material) by Extraction and Gravimetry" EPA-821-R-98-002, February 1999. Available at NTIS, PB-12949, U.S. Department of Commerce, 5285 Port Royal, Springfield, Virginia 22161.

    43 The application of clean techniques described in EPA's draft Method 1669: Sampling Ambient Water for Trace Metals at EPA Water Quality Criteria Levels (EPA-821-R-96-011) are recommended to preclude contamination at low-level, trace metal determinations.
  - 44 Available Cyanide, Method OIA-1677 (Available Cyanide by Flow Injection, Ligand Exchange, and Amperometry), ALPKEM, A Division of OI Analytical, P.O. Box 9010, College Station,

#### TABLE IC.—LIST OF APPROVED TEST PROCEDURES FOR NON-PESTICIDE ORGANIC COMPOUNDS

Parameter <sup>1</sup>	EPA method number <sup>27</sup>						
	GC	GC/MS	HPLC	Standard method 18th Ed.	ASTM	Other	
1. Acenaphthene	610	625, 1625	610	6410 B, 6440 B	D4657-92		
2. Acenaphthylene	610	625, 1625	610	6410 B, 6440 B	D4657-92		
3. Acrolein	603	<sup>4</sup> 604, 1624					
4. Acrylonitrile	603	4624, 1624	610				
5. Anthracene	610	625, 1625	610	6410 B, 6440 B	D4657-92		
6. Benzene	602	624, 1624		6210 B, 6220 B			
7. Benzidine		5 625, 1625	605			Note 3, p.1.	
8. Benzo(a)anthracene	610	625, 1625	610	6410 B, 6440 B	D4657-92		
9. Benzo(a)pyrene	610	625, 1625	610	6410 B, 6440 B	D4657-92		

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10. Benzo(b)fluoranthene	610	625, 1625	610	6410 B. 6440 B	D4657-92	
11. Benzo(g, h, i)perylene		625, 1625	610	6410 B. 6440 B	D4657-92	]
12. Benzo(k)fluoranthene		625, 1625		6410 B, 6440 B	D4657-92	<b>≥</b>
13. Benzyl chloride		020, 1020		01.02, 01.02	2 .00. 02	Note 3, p.130;
To. Bonzyi diliondo						Note 6, p.
14. Benzyl butyl phthalate	606	625, 1625		6410 B		Note 3, p.130: Note 6, p. S102.
15. Bis(2-chloroethoxy) methane		625, 1625		6410 B		₹
16. Bis(2-chloroethyl) ether	1	625, 1625		6410 B		<u>Ω</u>
17. Bis (2-ethylhexyl) phthalate		625, 1625		6410 B, 6230 B		70
18. Bromodichloromethane		624, 1624		6210 B, 6230 B		7
19. Bromoform		624, 1624		6210 B, 6230 B		<del>ĕ</del>
20. Bromomethane		624, 1624		6210 B, 6230 B		l %
21. 4-Bromophenylphenyl ether	'	625, 1625		6410 B		<b> </b>
22. Carbon tetrachloride		624, 1624		6230 B, 6410 B		Note 3, p.130.
23. 4-Chloro-3-methylphenol		625, 1625		6410 B, 6420 B		140te 3, p. 130.
24. Chlorobenzene	1	624, 1624		6210 B, 6220 B		Note 3, p.130.
24. Chioropenzene	001, 002	024, 1024		6230 B		Note 3, p. 130.
25. Chloroethane	601	624, 1624		6210 B, 6230 B		Note 3, p.130.  Note 3, p.130.  Agency
26. 2-Chloroethylvinyl ether		624, 1624		6210 B, 6230 B		i č
27. Chloraform		624, 1624		6210 B, 6230 B		Note, p.130.
28. Chloromethane		624, 1624		6210 B, 6230 B		140te, p. 130.
29. 2-Chloronaphthalene		625, 1625		6410 B		
30. 2-Chlorophenol		625, 1625		6410 B, 6420 B		
31. 4-Chlorophenylphenyl ether		625, 1625		6410 B		
32. Chrysene		625, 1625	610	6410 B. 6440 B	D4657-92	
33. Dibenzo(a,h)anthracene		625, 1625		6410 B, 6440 B	D4657-92	
34. Dibromochloromethane		624, 1624		6210 B, 6230 B	B-1007 02	
35. 1, 2-Dichlorobenzene		624,625,1625		6410 B, 6230 B, 6220 B		
36. 1, 3-Dichlorobenzene		624,625,1625		6410 B, 6230 B, 6220 B		
37. 1,4-Dichlorobenzene		624, 625, 1625		6410 B, 6220 B, 6230 B		
38. 3, 3-Dichlorobenzidine		625, 1625		6410 B		
39. Dichlorodifluoromethane				6230 B		
40. 1, 1-Dichloroethane		624, 1624		6230 B, 6210 B		
41. 1, 2-Dichloroethane		624, 1624		6230 B, 6210 B		
42. 1, 1-Dichloroethene		624, 1624		6230 B, 6210 B		
43. trans-1, 2-Dichloroethene		624, 1624		6230 B, 6210 B		
44. 2, 4-Dichlorophenol		625, 1625		6420 B, 6410 B		
45. 1, 2-Dichloropropane		624, 1624		6230 B, 6210 B		
46. cis-1, 3-Dichloropropene		624, 1624		6230 B, 6210 B		
47. trans-1, 3-Dichloropropene		624, 1624		6230 B, 6210 B		
48. Diethyl phthalate		625, 1625		6410 B		
49. 2, 4-Dimethylphenol		625, 1625		6420 B, 6410 B		
50. Dimethyl phthalate		625, 1625		6410 B		
51. Di-n-butyl phthalate		625, 1625		6410 B		1
52. Di-n-octyl phthalate		625, 1625		6410 B		w .
53. 2,4-Dinitrophenol		625, 1625		6420 B, 6410 B		136.3
54. 2,4-Dinitrotoluene	609	625, 1625		6410 B		<u> </u>
55. 2, 6-Dinitrotoluene						

TABLE IC.—LIST OF APPROVED TEST PROCEDURES FOR NON-PESTICIDE ORGANIC COMPOUNDS—Continued

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98. Pentachlorophenol	604	625, 1625		6410 B, 6630 B		Note 3, p. 140.
99. Phenanthrene	610	625, 1625	610	6410B, 6440B	D4657-87	
100. Phenol	604	625, 1625		6420B, 6410B		
101. Pyrene	610	625, 1625	610	6410B, 6440B	D4657-87	
102. 2,3,7,8-Tetrachlorodibenzofuran		1613				
103. 2,3,7,8-Tetrachlorodibenzo-p-dioxin		613, 1613 <sup>5</sup>				
104. 1,1,2,2-Tetrachloroethane	601	624, 1624		6230 B, 6210 B		Note 3, p. 130.
105. Tetrachloroethene	601	624, 1624		6230 B, 6410 B		Note 3, p. 130.
106. Toluene	602	624, 1624		6210B, 6220B		
107. 1,2,4-Trichlorobenzene	612	625, 1625		6410B		Note 3, p. 130.
108. 1,1,1-Trichloroethane	601	624, 1624		6210B, 6230B		
109. 1,1,2-Trichloroethane	601	624, 1624		6210B, 6230B		Note 3, p. 130.
110. Trichloroethene	601	624, 1624		6210 B, 6230 B		
111. Trichlorofluoromethane	601	624		6210 B, 6230 B		
112. 2,4,6-Trichlorophenol	604	625, 1625		6410B, 6240B		
113. Vinyl chloride	601	624, 1624		6210 B, 6230 B		

<sup>1</sup> All parameters are expressed in micrograms per liter (μg/L) except for Method 1613 in which the parameters are expressed in picograms per liter (pg/L).

The full text of Methods 601–613, 624, 625, 1624, and 1625, are given at appendix A, "Test Procedures for Analysis of Organic Pollutants," of this part 136. The full text of Method 1613 is incorporated by reference into this part 136 and is available from the National Technical Information Services as stock number PB95–104774. The standardized test procedure to be used to determine the method detection limit (MDL) for these test procedures is given at appendix B, "Definition and Procedures for the Determination of the Method Detection Limit," of this part

136.

3 "Methods for Benzidine: Chlorinated Organic Compounds, Pentachlorophenol and Pesticides in Water and Wastewater," U.S. Environmental Protection Agency, September, 1978.

4 Method 624 may be extended to screen samples for Acrolein and Acrylonitrile. However, when they are known to be present, the preferred method for these two compounds is Method 603 or Method 1624.

Method 625 may be extended to include benzidine, hexachlorocyclopentadiene, N-nitrosodimethylamine, and N-nitrosodiphenylamine. However, when they are known to be present, Methods 605, 607, and 612, or Method 1625, are preferred methods for these compounds.

Methods 605, 607, and 612, or method 1625, are preferred methods to mese compounds.

5a 625, Screening only,
6"Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency", Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981).

7 Each Analyst must make an initial, one-time demonstration of their ability to generate acceptable precision and accuracy with Methods 601–603, 624, 625, 1624, and 1625 (See Appendix A of this Part 136) in accordance with procedures each in section 8.2 of each of these Methods. Additionally, each laboratory, on an on-going basis must spike and analyze 10% (5% for Methods 624 and 625 and 100% for methods 1624 and 1625) of all samples to monitor and evaluate laboratory data quality in accordance with sections 8.3 and 8.4 of these Methods. When the recovery of any parameter falls outside the warning limits, the analytical results for that parameter in the unspiked sample are suspect and cannot be reported to demonstrate regulators.

NOTE: These warning limits are promulgated as an "interim final action with a request for comments."

<sup>8</sup> "Organochlorine Pesticides and PCBs in Wastewater Using Empore TM Disk", 3M Corporation Revised 10/28/94.

#### TABLE ID.—LIST OF APPROVED TEST PROCEDURES FOR PESTICIDES 1

Parameter	Method	EPA <sup>27</sup>	Standard methods 18th Ed.	ASTM	Other		
1. Aldrin	GC GC/MS		6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 8.		
2. Ametryn	GC TLC				Note 3, p. 83; Note 6, p. S68. Note 3, p. 94; Note 6, p. S16.		
4. Atraton	GC				Note 3, p. 83; Note 6, p. S68.		
5. Atrazine 6. Azinphos methyl	GC GC				Note 3, p. 83; Note 6, p. S68. Note 3, p. 25; Note 6, p. S51.		
7. Barban	TLC				Note 3, p. 104; Note 6, p. S64.		
8. α-BHC	GC GC/MS		6630 B & C	D3086–90	Note 3, p. 7; note 8.		
	I GC/IVIS	023	1 04 IU D		I		

TABLE ID.—LIST OF APPROVED TEST PROCEDURES FOR PESTICIDES 1—Continued

Parameter	Method	EPA <sup>27</sup>	Standard methods 18th Ed.	ASTM	Other
9. β-BHC	GC	608	6630 C	D3086-90	Note 8.
•	GC/MS	<sup>5</sup> 625			
0. δ-BHC		608		D3086-90	Note 8.
0. 0-Di 10	GC/MS	<sup>5</sup> 625			14016 0.
4 S DUO (Lindons)					Note 0 = 7: 4 = 00: 0
1. δ-BHC (Lindane)		608		D3086–90	Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	625	6410 B		
2. Captan	GC		6630 B	D3086-90	Note 3, p. 7.
3. Carbaryl	TLC				Note 3, p. 94: Note 6, p. S60.
4. Carbophenothion					Note 4, p. 30; Note 6, p. S73.
5. Chlordane			6630 B & C	D3086-90	
3. Officidatie	GC/MS	625			140te 5, p. 7, 110te 6.
	GC/IVIS	625	6410 В		
6. Chloropropham					Note 3, p. 104; Note 6, p. S64.
7. 2,4–D	GC		6640 B		Note 3, p. 115; Note 4, p. 35.
8. 4,4'-DDD	GC		6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 8.
- , ===	GC/MS		6410 B		, , , , , , , , , , , , , , , , , , , ,
9. 4.4′-DDE			6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 8.
9. 4,4 -DDL	GC/MS		6410 B	D3000-30	140te 5, p. 7, 110te 4, p. 50, 110te 6.
0 4 4′ DDT					Note 0 = 7: ==t= 4 = 00: ==t= 0
0. 4,4'-DDT		608	6630 B & C	D3086–90	Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	625	6410 B		
21. Demeton-O	GC				Note 3, p. 25; Note 6, p. S51.
2. Demeton-S	GC				Note 3, p. 25: Note 6, p. S51.
3. Diazinon					Note 3, p. 25; Note 4, p. 30; Note 6, p. S51
4. Dicamba					Note 3. p. 115.
5. Dichlofenthion					Note 4, p. 30; Note 6, p. S73.
6. Dichloran			6630 B & C		Note 3, p. 7.
7. Dicofol	GC			D3086–90	
8. Dieldrin	GC	608	6630 B & C		Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	625	6410 B		
9. Dioxathion	GC				Note 4, p. 30; Note 6, p. S73.
0. Disulfoton					Note 3, p. 25; Note 6, p. S51.
1. Diuron					Note 3, p. 104; Note 6, p. S64.
2. Endosulfan I			6630 B & C	D3086–90	Note 3, p. 7; note 8.
	GC/MS	<sup>5</sup> 625	6410 B		
3. Endosulfan II	GC	608	6630 B & C	D3086-90	Note 3, p. 7; note 8.
	GC/MS	<sup>5</sup> 625	6410 B		
4. Endosulfan Sulfate		608	6630 C		Note 8.
T. Eliocodiidii Callate	GC/MS	625	6410 B		11010 0.
F Fadria				D2000 00	Note 2 n 7: note 4 n 20: note 2
5. Endrin		608	6630 B & C	D3086–90	Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	<sup>5</sup> 625	6410 B		
6. Endrin aldehyde	GC	608			Note 8.
•	GC/MS	625			

	1		1	1	
37. Ethion	GC				Note 4, p. 30; Note 6, p. S73.
38. Fenuron	TLC				Note 3, p. 104; Note 6, p. S64.
39. Fenuron-TCA	TLC				Note 3, p. 104; Note 6, p. S64.
40. Heptachlor	GC	608	6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	625	6410 B		, , , , , , , , , , , , , , , , , , , ,
41. Heptachlor epoxide	GC	608	6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 6, p. S73; note
					8.
	GC/MS	625	6410 B		0.
42. Isodrin	GC				Note 4, p. 30; Note 6, p. S73.
43. Linuron	GC				Note 3, p. 104; Note 6, p. S64.
44. Malathion	GC		6630 C		Note 3, p. 25; Note 4, p. 30; Note 6, p. S51.
45. Methiocarb	TLC				Note 3, p. 94; Note 6, p. S60.
46. Methoxychlor	GC		6630 B & C	D3086-90	Note 3, p. 7; note 4, p. 30; note 8.
·					
47. Mexacarbate	TLC				Note 3, p. 94; Note 6, p. S60.
48. Mirex	GC		6630 B & C		Note 3, p. 7.
49. Monuron	TLC				Note 3, p. 104; Note 6, p. S64.
50. Monuron	TLC				Note 3, p. 104; Note 6, p. S64.
51. Nuburon	TLC				Note 3, p. 104; Note 6, p. S64.
52. Parathion methyl	GC		6630 C		Note 3, p. 25; Note 4, p. 30.
53. Parathion ethyl	GC		6630 C		Note 3, p. 25.
54. PCNB	GC		6630 B & C		Note 3, p. 7.
55. Perthane	GC			D3086-90	, p
56. Prometron	GC				Note 3, p. 83; Note 6, p. S68.
57. Prometryn	GC				Note 3, p. 83; Note 6, p. S68.
58. Propazine	GC				Note 3, p. 83; Note 6, p. S68.
59. Propham	TLC				Note 3, p. 104; Note 6, p. S64.
60. Propoxur	TLC				Note 3, p. 94; Note 6, p. S60.
61. Secbumeton	TLC				Note 3, p. 83; Note 6, p. S68.
62. Siduron	TLC				Note 3, p. 104; Note 6, p. S64.
63. Simazine	GC				Note 3, p. 83; Note 6, p. S68.
64. Strobane	GC		6630 B & C		
	TLC				Note 3, p. 7. Note 3, p. 104; Note 6, p. S64.
65. Swep	GC				
66. 2,4,5–T			6640 B		Note 3, p. 115; Note 4, p. 35.
67. 2,4,5-TP (Silvex)	GC		6640 B		Note 3, p. 115
68. Terbuthylazine	GC				Note 3, p. 83; Note 6, p. S68.
69. Toxaphene	GC	608		D3086–90	Note 3, p. 7; note 4, p. 30; note 8.
	GC/MS	625	6410 B		
70. Trifluralin	GC		6630 B		Note 3, p. 7.

Table ID notes:

1 Pesticides are listed in this table by common name for the convenience of the reader. Additional pesticides may be found under Table 1C, where entries are listed by chemical name.

2 The full text of Methods 608 and 625 are given at Appendix A. "Test Procedures for Analysis of Organic Pollutants," of this Part 136. The standardized test procedure to be used to determine the method detection limit (MDL) for these test procedures is given at Appendix B. "Definition and Procedure for the Determination of the Method Detection Limit", of this Part 136.

3"Methods for Benzidine, Chlorinated Organic Compounds, Pentachlorophenol and Pesticides in Water and Wastewater," U.S. Environmental Protection Agency, September, 1978. This EPA publication includes thin-layer chromatography (TLC) methods.

4"Methods for Analysis of Organic Substances in Water and Fluvial Sediments," Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 5, Chapter A3 (1987).

5 The method may be extended to include α-BHC, γ-BHC, endosulfan I, endosulfan II, and endrin. However, when they are known to exist, Method 608 is the preferred method.

6"Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency." Supplement to the Fifteenth Edition of Standard Methods for the Examination of Water and Wastewater (1981).

<sup>7</sup> Each analyst must make an initial, one-time, demonstration of their ability to generate acceptable precision and accuracy with Methods 608 and 625 (See Appendix A of this Part 136) in accordance with procedures given in section 8.2 of each of these methods. Additionally, each laboratory, on an-going basis, must spike and analyze 10% of all samples analyzed with Method 608 or 5% of all samples analyzed with Method 625 to monitor and evaluate laboratory data quality in accordance with Sections 8.3 and 8.4 of these methods. When the recovery of any parameter falls outside the warning limits, the analytical results for that parameter in the unspiked sample are suspect and cannot be reported to demonstrate regulatory compliance. These quality control requirements also apply to the Standard Methods, ASTM Methods, and other Methods cited.

NOTE: These warning limits are promulgated as an "Interim final action with a request for comments."

#### TABLE IE.—LIST OF APPROVED RADIOLOGIC TEST PROCEDURES

		Reference (method number or page)			
Parameter and units	Method	EPA <sup>1</sup>	Standard meth- ods 18th Ed.	ASTM	USGS <sup>2</sup>
Alpha-Total, pCi per liter     Alpha-Counting error, pCi per liter     Beta-Total, pCi per liter     Beta-Counting error, pCi     Beta-Counting error, pCi     (a) Radium Total pCi per liter     (b)Ra, pCi per liter	Proportional or scintillation counter Proportional or scintillation counter Proportional counter Proportional counter Proportional counter Scintillation counter	Appendix B	7110 B 7110 B 7110 B	D1943–90 D1943–90 D1890–90 D1890–90 D2460–90 D3454–91	pp. 75 and 78.3 P. 79. pp. 75 and 78.3 p. 79. p. 81.

Table IE notes:

<sup>&</sup>lt;sup>8</sup> "Organochlorine Pesticides and PCBs in Wastewater Using Empore<sup>TM</sup> Disk", 3M Corporation, Revised 10/28/94.

<sup>1</sup> Prescribed Procedures for Measurement of Radioactivity in Drinking Water," EPA-600/4-80-032 (1980), U.S. Environmental Protection Agency, August 1980.

<sup>&</sup>lt;sup>2</sup> Fishman, M.J. and Brown, Eugene," Selected Methods of the U.S. Geological Survey of Analysis of Wastewaters," U.S. Geological Survey, Open-File Report 76–177 (1976).

<sup>&</sup>lt;sup>3</sup> The method found on p. 75 measures only the dissolved portion while the method on p. 78 measures only the suspended portion. Therefore, the two results must be added to obtain the "total".

## **Environmental Protection Agency**

TABLE IF.—LIST OF APPROVED METHODS FOR PHARMACEUTICAL POLLUTANTS

Pharmaceuticals pollutants	CAS registry No.	Analytical method number
cetonitrile	. 75–05–8	1666/1671/D3371/D3695.
-amyl acetate	. 628–63–7	1666/D3695.
-amyl alcohol	. 71–41–0	1666/D3695
enzene	. 71–43–2	D4763/D3695/502.2/524.2.
-butyl-acetate	. 123–86–4	1666/D3695.
ert-butyl alcohol	. 75–65–0	1666.
hlorobenzene	. 108–90–7	502.2/524.2.
hloroform	. 67–66–3	502.2/524.2/551.
-dichlorobenzene	. 95–50–1	1625C/502.2/524.2.
,2-dichloroethane	. 107–06–2	D3695/502.2/524.2.
iethylamine	. 109–89–7	1666/1671.
imethyl sulfoxide	. 67–68–5	1666/1671.
thanol	. 64–17–5	1666/1671/D3695.
thyl acetate	. 141–78–6	1666/D3695.
-heptane	. 142–82–5	1666/D3695.
-hexane	. 110–54–3	1666/D3695.
obutyraldehyde	. 78–84–2	1666/1667.
sopropanol	. 67–63–0	1666/D3695.
sopropyl acetate	. 108–21–4	1666/D3695.
opropyl ether	. 108–20–3	1666/D3695.
nethanol	. 67–56–1	1666/1671/D3695.
Nethyl Cellosolve A	. 109–86–4	1666/1671
nethylene chloride	. 75–09–2	502.2/524.2
nethyl formate	. 107–31–3	1666.
-methyl-2-pentanone (MIBK)	. 108–10–1	1624C/1666/D3695/D4763/524.2.
henolhenol	. 108–95–2	D4763.
-propanol	. 71–23–8	1666/1671/D3695.
-propanone (acetone)	. 67–64–1	D3695/D4763/524.2.
etrahydrofuran		1666/524.2.
oluene	. 108–88–3	D3695/D4763/502.2/524.2.
riethlyamine	. 121–44–8	1666/1671.
ylenes	. (Note 1)	1624C/1666.

#### Table 1F note:

1. 1624C: m-xylene 108–38–3, o,p-xylene E–14095 (Not a CAS number; this is the number provided in the Environmental Monitoring Methods Index (EMMI) database.); 1666: m,p-xylene 136777–61–2, o-xylene 95–47–6.

(b) The full texts of the methods from the following references which are cited in Tables IA, IB, IC, ID, IE, and IF are incorporated by reference into this regulation and may be obtained from the sources identified. All costs cited are subject to change and must be verified from the indicated sources. The full texts of all the test procedures cited are available for inspection at the National Exposure Research Laboratory, Office of Research and Development, U.S. Environmental Protection Agency, 26 West Martin Luther King Dr., Cincinnati, OH 45268 and the Office of the Federal Register, 800 North Capitol Street, NW., Suite 700, Washington, DC.

# REFERENCES, SOURCES, COSTS, AND TABLE CITATIONS:

(1) The full texts of Methods 601–613, 624, 625, 1613, 1624, and 1625 are printed in appendix A of this part 136. The full text for determining the method detection limit when using the test proce-

dures is given in appendix B of this part 136. The full text of Method 200.7 is printed in appendix C of this part 136. Cited in: Table IB, Note 5; Table IC, Note 2; and Table ID, Note 2.

- (2) USEPA. 1978. Microbiological Methods for Monitoring the Environment, Water, and Wastes. Environmental Monitoring and Support Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. EPA/600/8–78/017. Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, Publ. No. PB–290329/AS. Cost: \$36.95. Table IA, Note 3.
- (3) "Methods for Chemical Analysis of Water and Wastes," U.S. Environmental Protection Agency, EPA-600/4-79-020, March 1979, or "Methods for Chemical Analysis of Water and Wastes," U.S. Environmental Protection Agency, EPA-600/4-79-020, Revised March 1983. Available from: ORD Publications, CERI, U.S. Environmental

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Protection Agency, Cincinnati, Ohio 45268, Table IB, Note 1.

- (4) "Methods for Benzidine, Chlorinated Organic Compounds, Pentachlorophenol and Pesticides in Water and Wastewater," U.S. Environmental Protection Agency, 1978. Available from: ORD Publications, CERI, U.S. Environmental Protection Agency, Cincinnati, Ohio 45268, Table IC, Note 3; Table D, Note 3.
- (5) "Prescribed Procedures for Measurement of Radioactivity in Drinking Water," U.S. Environmental Protection Agency, EPA-600/4-80-032, 1980. Available from: ORD Publications, CERI, U.S. Environmental Protection Agency, Cincinnati, Ohio 45268, Table IE, Note 1.
- (6) American Public Health Association. 1992. Standard Methods for the Examination of Water and Wastewater. 18th Edition. Amer. Publ. Hlth. Assoc., 1015 15th Street NW, Washington, DC 20005. Cost: \$160.00. Table IA, Note 4.
- (7) Ibid, 15th Edition, 1980. Table IB, Note 30; Table ID.
- (8) Ibid, 14th Edition, 1975. Table IB, Notes 17 and 27.
- (9) "Selected Analytical Methods Approved and Cited by the United States Environmental Protection Agency," Supplement to the 15th Edition of Standard Methods for the Examination of Water and Wastewater, 1981. Available from: American Public Health Association, 1015 Fifteenth Street NW., Washington, DC 20036. Cost available from publisher. Table IB, Note 10; Table IC, Note 6; Table ID, Note 6.
- (10) Annual Book of ASTM Standards, Water and Environmental Technology, Section 11, Volumes 11.01 and 11.02, 1994 in 40 CFR 136.3, Tables IB, IC, ID and IE.
- (11) USGS. 1989. U.S. Geological Survey Techniques of Water-Resources Investigations, Book 5, Laboratory Analysis, Chapter A4, Methods for Collection and Analysis of Aquatic Biological and Microbiological Samples, U.S. Geological Survey, U.S. Department of the Interior, Reston, Virginia. Available from: USGS Books and Open-File Reports Section, Federal Center, Box 25425, Denver, Colorado 80225. Cost: \$18.00. Table IA, Note 5.
- (12) "Methods for Determination of Inorganic Substances in Water and

Fluvial Sediments," by M.J. Fishman and Linda C. Friedman, Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 5 Chapter A1 (1989). Available from: U.S. Geological Survey, Denver Federal Center, Box 25425, Denver, CO 80225. Cost: \$108.75 (subject to change). Table IB, Note 2.

- (13) "Methods for Determination of Inorganic Substances in Water and Fluvial Sediments," N.W. Skougstad and others, editors. Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 5, Chapter A1 (1979). Available from: U.S. Geological Survey, Denver Federal Center, Box 25425, Denver, CO 80225. Cost: \$10.00 (subject to change), Table IB, Note 8.
- (14) "Methods for the Determination of Organic Substances in Water and Fluvial Sediments," Wershaw, R.L., et al, Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 5, Chapter A3 (1987). Available from: U.S. Geological Survey, Denver Federal Center, Box 25425, Denver, CO 80225. Cost: \$0.90 (subject to change). Table IB, Note 24; Table ID, Note 4.
- (15) "Water Temperature—Influential Factors, Field Measurement and Data Presentation," by H.H. Stevens, Jr., J. Ficke, and G.F. Smoot, Techniques of Water-Resources Investigations of the U.S. Geological Survey, Book 1, Chapter D1, 1975. Available from: U.S. Geological Survey, Denver Federal Center, Box 25425, Denver, CO 80225. Cost: \$1.60 (subject to change). Table IB, Note 32.
- (16) "Selected Methods of the U.S. Geological Survey of Analysis of Wastewaters," by M.J. Fishman and Eugene Brown; U.S. Geological Survey Open File Report 76–77 (1976). Available from: U.S. Geological Survey, Branch of Distribution, 1200 South Eads Street, Arlington, VA 22202. Cost: \$13.50 (subject to change). Table IE, Note 2.
- (17) "Official Methods of Analysis of the Association of Official Analytical Chemicals", Methods manual, 15th Edition (1990). Price: \$240.00. Available from: The Association of Official Analytical Chemists, 2200 Wilson Boulevard, Suite 400, Arlington, VA 22201. Table IB, Note 3.
- (18) "American National Standard on Photographic Processing Effluents," April 2, 1975. Available from: American

National Standards Institute, 1430 Broadway, New York, New York 10018. Table IB, Note 9.

- (19) "An Investigation of Improved Procedures for Measurement of Mill Effluent and Receiving Water Color," NCASI Technical Bulletin No. 253, December 1971. Available from: National Council of the Paper Industry for Air and Stream Improvements, Inc., 260 Madison Avenue, New York, NY 10016. Cost available from publisher. Table IB, Note 18.
- (20) Ammonia, Automated Electrode Method, Industrial Method Number 379–75WE, dated February 19, 1976. Technicon Auto Analyzer II. Method and price available from Technicon Industrial Systems, Tarrytown, New York 10591. Table IB, Note 7.
- (21) Chemical Oxygen Demand, Method 8000, Hach Handbook of Water Analysis, 1979. Method price available from Hach Chemical Company, P.O. Box 389, Loveland, Colorado 80537. Table IB, Note 14.
- (22) OIC Chemical Oxygen Demand Method, 1978. Method and price available from Oceanography International Corporation, 512 West Loop, P.O. Box 2980, College Station, Texas 77840. Table IB, Note 13.
- (23) ORION Research Instruction Manual, Residual Chlorine Electrode Model 97–70, 1977. Method and price available from ORION Research Incorporation, 840 Memorial Drive, Cambridge, Massachusetts 02138. Table IB, Note 16.
- (24) Bicinchoninate Method for Copper. Method 8506, Hach Handbook of Water Analysis, 1979, Method and price available from Hach Chemical Company, P.O. Box 300, Loveland, Colorado 80537. Table IB, Note 19.
- (25) Hydrogen Ion (pH) Automated Electrode Method, Industrial Method Number 378-75WA. October 1976. Bran & Luebbe (Technicon) Auto Analyzer II. Method and price available from Bran & Luebbe Analyzing Technologies, Inc. Elmsford, N.Y. 10523. Table IB, Note 21.
- (26) 1,10-Phenanthroline Method using FerroVer Iron Reagent for Water, Hach Method 8008, 1980. Method and price available from Hach Chemical Company, P.O. Box 389 Loveland, Colorado 80537. Table IB, Note 22.

- (27) Periodate Oxidation Method for Manganese, Method 8034, Hach Handbook for Water Analysis, 1979. Method and price available from Hach Chemical Company, P.O. Box 389, Loveland, Colorado 80537. Table IB, Note 23.
- (28) Nitrogen, Nitrite—Low Range, Diazotization Method for Water and Wastewater, Hach Method 8507, 1979. Method and price available from Hach Chemical Company, P.O. Box 389, Loveland, Colorado 80537. Table IB, Note 25.
- (29) Zincon Method for Zinc, Method 8009. Hach Handbook for Water Analysis, 1979. Method and price available from Hach Chemical Company, P.O. Box 389, Loveland, Colorado 80537. Table IB, Note 33.
- (30) "Direct Determination of Elemental Phosphorus by Gas-Liquid Chromatography," by R.F. Addison and R.G. Ackman, Journal of Chromatography, Volume 47, No. 3, pp. 421–426, 1970. Available in most public libraries. Back volumes of the Journal of Chromatography are available from Elsevier/North-Holland, Inc., Journal Information Centre, 52 Vanderbilt Avenue, New York, NY 10164. Cost available from publisher. Table IB, Note 28.
- (31) "Direct Current Plasma (DCP) Optical Emission Spectrometric Method for Trace Elemental Analysis of Water and Wastes", Method AES 0029, 1986-Revised 1991, Fison Instruments, Inc., 32 Commerce Center, Cherry Hill Drive, Danvers, MA 01923. Table B, Note 34.
- (32) "Closed Vessel Microwave Digestion of Wastewater Samples for Determination of Metals, CEM Corporation, P.O. Box 200, Matthews, North Carolina 28106–0200, April 16, 1992. Available from the CEM Corporation. Table IB, Note 36.
- (33) "Organochlorine Pesticides and PCBs in Wastewater Using Empore TM Disk" Test Method 3M 0222, Revised 10/28/94. 3M Corporation, 3M Center Building 220–9E–10, St. Paul, MN 55144–1000. Method available from 3M Corporation. Table IC, Note 8 and Table ID, Note 8.
- (34) USEPA. 1993. Methods for Measuring the Acute Toxicity of Effluents to Freshwater and Marine Organisms.

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Fourth Edition, December 1993. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio (EPA/600/4–90/027F). Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, Publ. No. PB-91-167650. Cost: \$31.00. Table IA, Note 17. See changes in the manual, listed in Part V of this rule.

(35) "Nitrogen, Total Kjeldahl, Method PAI-DK01 (Block Digestion, Steam Distillation, Titrimetric Detection)", revised 12/22/94. Available from Perstorp Analytical Corporation, 9445 SW Ridder Rd., Suite 310, P.O. Box 648, Wilsonville, OK 97070. Table IB, Note 39.

(36) "Nitrogen, Total Kjeldahl, Method PAI-DK02 (Block Digestion, Steam Distillation, Colorimetric Detection)", revised 12/22/94. Available from Perstorp Analytical Corporation, 9445 SW Ridder Rd., Suite 310, P.O. Box 648, Wilsonville, OK 97070. Table IB, Note 40.

(37) "Nitrogen, Total Kjeldahl, Method PAI-DK03 (Block Digestion, Automated FIA Gas Diffusion)", revised 12/22/94. Available from Perstorp Analytical Corporation, 9445 SW Ridder Rd., Suite 310, P.O. Box 648, Wilsonville, OK 97070. Table IB, Note 41.

(38) USEPA. 1994. Short-term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Freshwater Organisms. Third Edition. July 1994. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. (EPA/600/4-91/002). Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, Publ. No. PB-92-139492. Cost: \$31.00. Table IA, Note 8.

(39) USEPA. 1994. Short-term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Marine and Estuarine Organisms. Second Edition, July 1994. Environmental Monitoring Systems Laboratory, U.S. Environmental Protection Agency, Cincinnati, Ohio. EPA/600/4-91/003. Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161, Publ.

No. PB-92-139484. Cost: \$45.00. Table IA, Note 9.

(40) EPA Methods 1666, 1667, and 1671 listed in the table above are published in the compendium titled Analytical Methods for the Determination of Pollutants in Pharmaceutical Manufacturing Industry Wastewaters (EPA 821-B-98-016). EPA Methods 502.2 and 524.2 have been incorporated by reference into 40 CFR 141.24 and are in Methods for the Determination of Organic Compounds in Drinking Water, EPA-600/4-88-039, December 1988, Revised, July 1991, and Methods for the Determination of Organic Compounds in Drinking Water-Supplement II, EPA-600/R-92-129, August 1992, respectively. These EPA test method compendia are available from the National Technical Information Service, NTIS PB91-231480 and PB92-207703, U.S. Department of Commerce, 5285 Port Royal Road, Springfield, Virginia 22161. The toll-free number is 800-553-6847. ASTM test methods D3371, D3695, and D4763 are available from the American Society for Testing and Materials, 100 Barr Harbor Drive, West Conshohocken, PA 19428-2959.

(41) USEPA. 2001. Method 1631, Revision C, "Mercury in Water by Oxidation, Purge and Trap, and Cold Vapor Atomic Fluorescence Spectrometry." March 2001. Office of Water, U.S. Environmental Protection Agency (EPA-821-R-01-024). Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161. Publication No. PB2001-102796. Cost: \$25.50. Table IB, Note 43.

(42) USEPA, January 1999 Errata for the Effluent and Receiving Water Testing Manuals: Acute Toxicity Effluents and Receiving Waters to Freshwater and Marine Organisms; Short-Term Methods for Estimating the Chronic Toxicity of Effluents and Receiving Waters to Freshwater Organisms; and Short-Term Methods for Estimating the Chronic Toxicity Effluents and Receiving Waters to Marine and Estuarine Organisms. U.S. Environmental Protection Agency, Office of Research and Development, Duluth, MN. EPA-600/R-98/182.

(43) Method OIA-1677, Available Cyanide by Flow Injection, Ligand Exchange, and Amperometry. August 1999. ALPKEM, OI Analytical, Box 648,

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Wilsonville, Oregon 97070 (EPA-821-R-99-013). Available from: National Technical Information Service, 5285 Port Royal Road, Springfield, Virginia 22161. Publication No. PB99-132011. Cost: \$22.50. Table IB, Note 44.

(c) Under certain circumstances the Regional Administrator or the Director in the Region or State where the discharge will occur may determine for a particular discharge that additional parameters or pollutants must be reported. Under such circumstances, additional test procedures for analysis of pollutants may be specified by the Regional Administrator, or the Director upon the recommendation of the Director of the Environmental Monitoring Systems Laboratory—Cincinnati.

(d) Under certain circumstances, the Administrator may approve, upon recommendation by the Director, Environmental Monitoring Systems Laboratory—Cincinnati, additional alternate test procedures for nationwide use.

(e) Sample preservation procedures, container materials, and maximum allowable holding times for parameters

cited in Tables IA, IB, IC, ID, and IE are prescribed in Table II. Any person may apply for a variance from the prescribed preservation techniques, container materials, and maximum holding times applicable to samples taken from a specific discharge. Applications for variances may be made by letters to the Regional Administrator in the Region in which the discharge will occur. Sufficient data should be provided to assure such variance does not adversely affect the integrity of the sample. Such data will be forwarded, by the Regional Administrator, to the Director of the Environmental Moni-Systems Laboratory—Cintoring cinnati, Ohio for technical review and recommendations for action on the variance application. Upon receipt of the recommendations from the Director of the Environmental Monitoring Systems Laboratory, the Regional Administrator may grant a variance applicable to the specific charge to the applicant. A decision to approve or deny a variance will be made within 90 days of receipt of the application by the Regional Administrator.

TABLE II—REQUIRED CONTAINERS, PRESERVATION TECHNIQUES, AND HOLDING TIMES

Parameter No./name	Container 1	Preservation 2, 3	Maximum holding time 4
Table IA—Bacteria Tests:			
1-4 Coliform, fecal and total	P,G	Cool, 4C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	6 hours.
5 Fecal streptococci	P,G	Cool, 4C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 5	6 hours.
Table IA—Aquatic Toxicity Tests:			
6-10 Toxicity, acute and chronic	P,G	Cool, 4 °C 16	36 hours.
Table IB—Inorganic Tests:	1	ı	ı
1. Acidity	P, G	Cool, 4°C	14 days.
2. Alkalinity	P, G	do	Do.
4. Ammonia		Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.
9. Biochemical oxygen demand		Cool, 4°C	48 hours.
10. Boron		HNO <sub>3</sub> TO pH<2	6 months.
	Quartz.		
11. Bromide	P, G	None required	28 days.
14. Biochemical oxygen demand, carbonaceous	P, G	Cool, 4°C	48 hours.
15. Chemical oxygen demand	P, G	Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.
16. Chloride		None required	Do.
17. Chlorine, total residual	P, G	do	Analyze immediately.
21. Color		Cool, 4°C	48 hours.
23-24. Cyanide, total and amenable to	P, G	Cool, 4°C, NaOH to pH>12,	14 days.6
chlorination.		0.6g ascorbic acid 5.	
25. Fluoride		None required	28 days.
27. Hardness		HNO <sub>3</sub> to pH<2, H <sub>2</sub> SO <sub>4</sub> to pH<2	6 months.
28. Hydrogen ion (pH)		None required	Analyze immediately.
31, 43. Kjeldahl and organic nitrogen	P, G	Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.
Metals:7			
18. Chromium VI			24 hours.
35. Mercury			
3, 5–8, 12, 13, 19, 20, 22, 26, 29, 30, 32–34, 36,	P, G	do	6 months.
37, 45, 47, 51, 52, 58–60, 62, 63, 70–72, 74,			
75. Metals, except boron, chromium VI and			
mercury.			
38. Nitrate			
39. Nitrate-nitrite	I P, G	Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.

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TABLE II—REQUIRED CONTAINERS, PRESERVATION TECHNIQUES, AND HOLDING TIMES—Continued

Parameter No./name	Container 1	Preservation 2, 3	Maximum holding time 4
40. Nitrite	P, G	Cool, 4°C	48 hours.
41. Oil and grease	G	Cool to 4°C, HCl or H <sub>2</sub> SO <sub>4</sub> to pH<2.	28 days.
42. Organic Carbon	P, G	Cool to 4 °C HC1 or H <sub>2</sub> SO4 or H <sub>3</sub> PO4, to pH<2.	28 days.
44. Orthophosphate	P, G	Filter immediately, Cool, 4°C	48 hours.
46. Oxygen, Dissolved Probe	G Bottle and top.	None required	Analyze immediately.
47. Winkler	do	Fix on site and store in dark	8 hours.
48. Phenols	G only	Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.
49. Phosphorus (elemental)	G	Cool, 4°C	48 hours.
50. Phosphorus, total	P, G	Cool, 4°C, H <sub>2</sub> SO <sub>4</sub> to pH<2	28 days.
53. Residue, total	P, G	Cool, 4°C	7 days.
54. Residue, Filterable	P, G	do	7 days.
55. Residue, Nonfilterable (TSS)	P, G	do	7 days.
56. Residue, Settleable	P, G	do	48 hours.
57. Residue, volatile	P, G	do	7 days.
61. Silica	P, PFTE, or Quartz.	Cool, 4 °C	28 days.
64. Specific conductance	P, G	do	Do.
65. Sulfate	P, G	do	Do.
66. Sulfide	P, G	Cool, 4°C add zinc acetate plus sodium hydroxide to pH>9.	7 days.
67. Sulfite	P, G	None required	Analyze immediately.
68. Surfactants	P ,G	Cool, 4°C	48 hours.
69. Temperature	P, G	None required	Analyze.
73. Turbidity	P, G	Cool, 4°C	48 hours.
Table IC—Organic Tests <sup>8</sup>			
13, 18–20, 22, 24–28, 34–37, 39–43, 45–47, 56, 76, 104, 105, 108–111, 113. Purgeable Halocarbons.	G, Teflon- lined sep- tum.	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup> .	14 days.
6, 57, 106. Purgeable aromatic hydrocarbons	do	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> , <sup>5</sup> HCl to pH2 <sup>9</sup> .	Do.
3, 4. Acrolein and acrylonitrile	do	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> ,5 adjust pH to 4–5 <sup>10</sup> .	Do.
23, 30, 44, 49, 53, 77, 80, 81, 98, 100, 112. Phenols <sup>11</sup> .	G, Teflon- lined cap	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	7 days until extraction; 40 days after extraction.
7, 38. Benzidines 11	do	do	7 days until extraction.13
14, 17, 48, 50–52. Phthalate esters 11	do	Cool, 4 °C	7 days until extraction; 40 days after extrac-
92 94 Nitrocominos II I4	4-	Cool 4 %C 0 0000/ No C 0 5	tion.
82–84. Nitrosamines 11 14	do	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> , <sup>5</sup> store in dark.	Do.
88–94. PCBs 11	do	Cool, 4 °C	Do.
54, 55, 75, 79. Nitroaromatics and isophorone 11	do	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> , <sup>5</sup> store in dark.	Do.
1, 2, 5, 8–12, 32, 33, 58, 59, 74, 78, 99, 101. Polynuclear aromatic hydrocarbons <sup>11</sup> .	do	do	Do.
15, 16, 21, 31, 87. Haloethers 11	do	Cool, 4 °C, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup>	Do.
29, 35-37, 63-65, 73, 107. Chlorinated hydro-	do	Cool, 4 °C	Do.
carbons <sup>11</sup> . 60–62, 66–72, 85, 86, 95–97, 102, 103. CDDs/			
CDFs <sup>11</sup> .			
aqueous: field and lab preservation	G	Cool, 0–4 °C, pH<9, 0.008% Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> <sup>5</sup> .	1 year.
Solids, mixed phase, and tissue: field preservation	do	Cool, <4 °C	7 days.
Solids, mixed phase, and tissue: lab preservation Table ID—Pesticides Tests:	do	Freeze, <-10 °C	1 year.
1–70. Pesticides <sup>11</sup> Table IE—Radiological Tests:	do	Cool, 4°C, pH 5–9 15	Do.
1-5. Alpha, beta and radium	P, G	HNO <sub>3</sub> to pH<2	6 months.

Table II Notes

1 Polyethylene (P) or glass (G). For microbiology, plastic sample containers must be made of sterilizable materials (polypropylene or other autoclavable plastic).

2 Sample preservation should be performed immediately upon sample collection. For composite chemical samples each aliquot should be preserved at the time of collection. When use of an automated sampler makes it impossible to preserve each aliquot, then chemical samples may be preserved by maintaining at 4°C until compositing and sample splitting is completed.

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³When any sample is to be shipped by common carrier or sent through the United States Mails, it must comply with the Department of Transportation Hazardous Materials Regulations (49 CFR part 172). The person offering such material for transportation is responsible for ensuring such compliance. For the preservation requirements of Table II, the Office of Hazardous Materials and the properties of the preservation requirements of Table II, the Office of Hazardous Materials Hydrochloric acid (HCI) in water solutions at concentrations of 0.04% by weight or less (pH about 1.96 or greater); Nitric acid (HNO<sub>3</sub>) in water solutions at concentrations of 0.15% by weight or less (pH about 1.62 or greater); Sulfuric acid (H<sub>2</sub>SO<sub>3</sub>) in water solutions at concentrations of 0.35% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hydroxide (NaOH) in water solutions at concentrations of 0.080% by weight or less (pH about 1.15 or greater); and Sodium hyd

The pri adjustment is not required in actionin will not be measured, samples for acrolein receiving no pri adjustment must be analyzed within 3 days of sampling.

11 When the extractable analytes of concern fall within a single chemical category, the specified preservative and maximum holding times should be observed for optimum safeguard of sample integrity. When the analytes of concern fall within two or more chemical categories, the sample may be preserved by cooling to 4°C, reducing residual chlorine with 0.008% sodium thiosulfate, storing in the dark, and adjusting the pH to 6–9; samples preserved in this manner may be held for seven days before extraction and for forty days after extraction. Exceptions to this optional preservation and holding time procedure are noted in footnote 5 (re the requirement for thiosulfate reduction of residual chlorine), and footnotes 12, 13 (re the analysis of benzi-

12 If 1,2-diphenylhydrazine is likely to be present, adjust the pH of the sample to 4.0±0.2 to prevent rearrangement to benzi-

dine.

13 Extracts may be stored up to 7 days before analysis if storage is conducted under an inert (oxidant-free) atmosphere.

14 For the analysis of diphenylnitrosamine, add 0.008% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and adjust pH to 7–10 with NaOH within 24 hours of sam-

pling.

15 The pH adjustment may be performed upon receipt at the laboratory and may be omitted if the samples are extracted within 72 hours of collection. For the analysis of aldrin, add 0.008% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>.

To Sufficient ice should be placed with the samples in the shipping container to ensure that ice is still present when the samples arrive at the laboratory. However, even if ice is present when the samples arrive, it is necessary to immediately measure the temperature of the samples and confirm that the 4C temperature maximum has not been exceeded. In the isolated cases where it can be documented that this holding temperature can not be met, the permittee can be given the option of on-site testing or can request a variance. The request for a variance should include supportive data which show that the toxicity of the effluent samples is not reduced because of the increased holding temperature.

138 FR 28758, Oct. 16, 1973, as amended at 41 FR 52781, Dec. 1, 1976; 49 FR 43251, 43258, 43259. Oct. 26, 1984; 50 FR 691, 692, 695, Jan. 4, 1985; 51 FR 23693, June 30, 1986; 52 FR 33543, Sept. 3, 1987; 55 FR 24534, June 15, 1990; 55 FR 33440, Aug. 15, 1990; 56 FR 50759, Oct. 8, 1991; 57 FR 41833, Sept. 11, 1992; 58 FR 4505, Jan. 31, 1994; 60 FR 17160, Apr. 4, 1995; 60 FR 39588, 39590, Aug. 2, 1995; 60 FR 44672, Aug. 28, 1995; 60 FR 53542, 53543, Oct. 16, 1995; 62 FR 48403, 48404, Sept. 15, 1997; 63 FR 50423, Sept. 21, 1998; 64 FR 4978, Feb. 2, 1999; 64 FR 10392, Mar. 4, 1999; 64 FR 26327, May 14, 1999; 64 FR 30433, 30434, June 8, 1999; 64 FR 73423, Dec. 30, 1999; 66 FR 32776, June 18, 2001]

#### § 136.4 Application for alternate test procedures.

- (a) Any person may apply to the Regional Administrator in the Region where the discharge occurs for approval of an alternative test procedure.
- (b) When the discharge for which an alternative test procedure is proposed occurs within a State having a permit program approved pursuant to section 402 of the Act, the applicant shall submit his application to the Regional Administrator through the Director of the State agency having responsibility for issuance of NPDES permits within such State.
- (c) Unless and until printed application forms are made available, an application for an alternate test procedure may be made by letter in trip-

licate. Any application for an alternate test procedure under this paragraph (c)

- (1) Provide the name and address of the responsible person or firm making the discharge (if not the applicant) and the applicable ID number of the existing or pending permit, issuing agency, and type of permit for which the alternate test procedure is requested, and the discharge serial number.
- (2) Identify the pollutant or parameter for which approval of an alternate testing procedure is being requested.
- (3) Provide justification for using testing procedures other than those specified in Table I.
- (4) Provide a detailed description of the proposed alternate test procedure, together with references to published